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An Activity Series - Judy Chen

The activity series is a list of the metals in order of ease of reaction, or the ability of the element to give up electrons. Metals will give up electrons to become positively charged ions, called cations. When an atom

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loses its electrons, it is oxidized. Nonmetals will accept electrons to become negatively charged ions, called anions.

Activity Series Lab - Scribd

1. Oxidation is the process of losing electrons. 2. Reduction is the process of gaining electrons. 3. A good oxidizing agent is easily reduced because it can easily gain

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electrons from the reducing agent. 4. a) No, the less reactive metal, Pt, should stay solid, but it is

Developing the Activity Series

An Activity Series of
Ions Lab Report Shakil
M, Horby L, Shaquile P,
Fazal M April 11th 2012

Mr. Vu Chemistry

Introduction: Many
reactions occur in our
daily lives to benefit
and improve our

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wellness and
surroundings.

Activity Series Lab Report by Darbi Cook on Prezi

1 Most metals have.
high and
conductivities, luster,
malleability, and have
low low luster,
malleability, and tñtw.
additiort, relative
metals have ionization

**home [intro.chem.ok
state.edu]**

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before performing this lab. Goal/Objective: The students will develop an activity series based on their own lab observations and will use it to predict and explain single replacement reactions and oxidation-reduction. Here is the website where all of the NACE corrosion labs can be found.

**ACTIVITY SERIES
LAB - Auburn School**
Page 8/24

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The hypothesis has been proven after the experiment. The activity series were correctly found for the metals Magnesium, Zinc, Lead, and Copper and, the halogens Chlorine, Bromine, and Iodine by performing a series of reactions. This lab has almost a hundred percent accuracy since all the data found had matched the actual

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AP Chemistry Lab 3 1 Activity Series of Metals and Nonmetals

The metal activity series indicates the reactivity of metals and how strongly they are reduced. The series begins with lithium, potassium, calcium, and sodium. Chemistry . Science ... Answer. Yes. Magnesium is above copper on the

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reactivity series of metals. Therefore, it will replace the copper in the copper chloride, producing magnesium ...

Chemistry Lab Report (The Activity Series) - Sarah Jackson

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Lab. Uploaded by
Jonathan_Khan7.

Description: ... The God
Particle: If the Universe
Is the Answer, What Is
the Question? Physics II
For Dummies. Organic
Chemistry II For
Dummies. Beyond
Weird: Why Everything
You ...

**Activity Series -
yaksic.com - high
school chemistry lab**

...

Students can add their

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own graphs here.

Activity Series The purpose of this lab was to observe the reactions of metals and the importance of the activity series.

Magnesium-HCL Zinc-HCL Iron-HCL Tin-HCL Copper-HCL Copper-AgNO₃ Grade and Comments Class 3. .

Observe the wells for a

Solved: REPORT SHEET EXPERIMENT Activity Series 10 A.

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An Activity Series Lab by av s on Prezi

Virtual Lab: Activity Series. ... This is the activity series which lists the most reactive metals on top and the least reactive on the bottom. Compare your answer to question #3 with the activity series. Are your results in the same order? Why?
(Click [here](#) for Table J)

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EXPERIMENT 8
Activity Series
(Single
Displacement
Reactions)

Answers to questions
in complete sentences

. REVISED 12/2003

Activity Series Lab -
Observations for Part 1

KNO_3 $\text{Mg}(\text{NO}_3)_2$

$\text{Zn}(\text{NO}_3)_2$ CuSO_4

AgNO_3 Distilled H_2O

Copper Iron

Magnesium Tin Zinc .

REVISED 12/2003

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Activity Series Lab -
Observations for Part 2
Reaction with HCl
Copper

activity series lab - Kris Kelly Activity Series Lab CHEM ...

AP Chemistry Lab 3 1
Activity Series of
Metals and Nonmetals
PURPOSE To determine
an activity series for
metals and an activity
series for nonmetals.
INTRODUCTION In this
experiment you will

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study some metals and some nonmetals to find their relative reactivity. A ranking according to reactivity is called an activity series.

An Activity Series Inquiry Guidance and AP* Chemistry

...

The activity series is a list of elements in decreasing order of their reactivity. Since metals replace other metals, while

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nonmetals replace other nonmetals, they each have a separate activity series. The table $\{\{1\}\}$ below is an activity series of most common metals, and the table $\{\{2\}\}$ is an activity series of the ...

Virtual Lab Activity Series - Mr. Palermo's Flipped ...
Activity Series lab experiment. Activity Series . Let's become

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familiar with the relative activities of metals in chemicals! You need this equipment: 7 small test tubes, test tube rack.

Activity Series Lab Answers |

SchoolWorkHelper

EXPERIMENT 8: Activity Series (Single Displacement Reactions) Name: _____

Pre- laboratory

Questions and

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Exercises Due before lab begins. Answer in separate sheet of paper. 1. List the three metals, Mg, Pb, Cu, Zn, and Fe in order of increasing activity (least to highest).

An Activity Series Lab Answers

The activity series is a list of metals in order from most active to least active from top to bottom; therefore, for a

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single replacement reaction to occur, the lone metal should be higher in the activity series than the metal in the compound for a reaction to take place.

Metal Activity Series - Chemistry | Socratic

Lab Furniture Lab
Techniques ... An
Activity Series Inquiry
Guidance and AP*
Chemistry Curriculum
Alignment Transition

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Guide; An Activity Series Inquiry Guidance and AP* Chemistry Curriculum Alignment Transition Guide. DC11214. Price: FREE. Learn more about downloading digital content.

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Answer to REPORT SHEET EXPERIMENT Activity Series 10 A. Reactions of Metals with Acid Reaction Metal with HCl

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An Activity Series of Ions Lab | Chemical Reactions ...

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